# **Basic classes of biomolecules**



## Nucleotide polymerization

*Regioselective formation of 3'-5' phosphodiester bonds between nucleotides* 



# Nucleosides - nucleobases + sugars



н

н

н

Purines

nucleobases, followed by regio- and diastereoselective glycosidation  $\rightarrow$  monomeric nucleosides

#### **RNA oligomers – prebiotic disconnections**



#### *Literature sources*

J. Oro, Biochem. Biophys. Res. Commun. **1960**, *2*, 407–412.

J. D. Sutherland, *Cold Spring Harbor Perspect. Biol.* **2010**, 2, a005439.

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J. D. Sutherland, et al. Nat. Chem. 2013, 5, 383–389.

L. E. Orgel, Crit. Rev. Biochem. Mol. Biol. 2004, 39, 99-123.

Powner, M. W.; Sutherland, J. D.; Szostak, J. W. J. Am. Chem. Soc. 2010, 132, 16677

T. Carell, Nature 2016, 352(6287), 833-836

# Prebiotic synthesis of nucleobases





Guanine

Pyrimidines



#### Prebiotic synthesis of adenine – the pentamer of HCN!



1960 - Oró's synthesis of adenine 2 from hydrogen cyanide 1 and ammonia (general acid–base catalysis, presumed to operate in most steps, is only shown once).
Heating ammonium cyanide at 70°C for a few days
→ 0.5% adenine
Heating HCN with liquid ammonia in a sealed tube → 20% adenine

The photochemical shortcut discovered by Ferris and Orgel is shown by the red arrow.

Optimized yields – up to 20% for adenine, 3% for guanine

Eutectic freezing (-20<sup>o</sup>C) increases the yield of DAMN formation by concentrating HCN between pure ice crystals



J. Oro Biochem. Biophys. Res. Commun. 1960, 2, 407.

J. P. Ferris, L. E. Orgel, J. Am. Chem. Soc. 1966, 88, 1074



# **Prebiotic synthesis of pyrimidines**

**Cyanoacetylene** is a major product of electric discharges in the mixture of nitrogen and methane **Cyanoacetylene** can be hydrolysed to cyanoacetaldehyde. That compound can condense with urea to form cytosine



**Cyanoacetylene** incubated with saturated solution of urea yields up to 50% **cytosine**. Other methods typically yield up to 5% cytosine. It is further converted to uracil by hydrolysis. *R. Shapiro PNAS* **1999**, *96*, 4396-4401

# Prebiotic synthesis of nucleosides, nucleotides and RNA



#### **Prebiotic synthesis of nucleosides**



The difficulties of assembling beta-ribonucleosides by nucleobase ribosylation:

The many different forms of ribose **3** adopted in aqueous solution. The pyranose (p) and furanose (f) forms interconvert via the open-chain aldehyde (a), which is also in equilibrium with an open-chain aldehyde hydrate (not shown).

#### **Prebiotic synthesis of nucleosides**



Heating purines with ribose and magnesium salts yields beta-nucleosides, although with low yields
 β-inosine – 8%, β-adenosine – 4%, β-guanosine – 9%
 Other isomers (e.g. alpha-glycosides) also present.

W. Fuller, R. Sanchez, L. Orgel J. Mol. Biol. 1972, 67, 25-33



Adenine tautomerism and the ribosylation step necessary to make the adenosine **11** thought to be needed for RNA assembly. The low abundance of the reactive entities **13** and **14** is partly responsible for the low yield of **11**. The reason for the lower nucleophilicity of *N1* of the pyrimidines, and the conventional synthetic chemist's solution to the problems of ribosylation.

# Prebiotic synthesis of pyrimidine nucleosides



No direct synthesis of pyrimidine nucleosides from ribose reported so far.

# **RNA** oligomers – prebiotic disconnections



## **Prebiotic synthesis of nucleosides**



Ara-3'P + NH<sub>2</sub>CN + HCC-CN  $\rightarrow$  Cyt-2',3'cP: A. Ingar, R. W. A. Luke, B. R. Hayter, J. D. Sutherland *ChemBioChem* **2003**, *4*, 504-507

#### **Prebiotic synthesis of oligonucleotides**



Activated ribonucleotides in the potentially prebiotic assembly of RNA. Potential P–O bond forming polymerization chemistry is indicated by the curved arrows.

# Cyanosulfidic chemistry



# **RNA** oligomers – prebiotic disconnections





#### Cyanosulfidic chemistry for the Kiliani-Fischer homologation

# Cyanosulfidic chemistry

Cyanamide (8) is a fertilizer from calcium carbide and nitrogen. The calcium carbide is formed upon heating of calcium cyanide

H2N

Cyanoacetylene (7) is a major nitrogen-containing product of the action of an electric discharge on a mixture of methane and nitrogen.

M. W. Powner, B. Gerland, J. D. Sutherland, Nature 2009, 459, 239-242





# Cyanosulfidic chemistry

The recently uncovered route to activated pyrimidine nucleotides 2.

The nucleobase ribosylation problem is circumvented by the assembly proceeding through 2-aminooxazole **21**, which can be thought of as the chimera of half a pentose sugar and half a nucleobase. The second half of the pentose - glyceraldehyde **5** -and the second half of the nucleobase—cyanoacetylene **7**—are then added sequentially to give the anhydronucleoside **23**.

Phosphorylation and rearrangement of **23** then furnishes **2** (B=C), and UV irradiation effects the partial conversion of **2** (B=C) to **2** (B=U).



#### Cytosine-2',3'cP – step 1: 2-amino-oxazole

#### Cytosine-2',3'cP – step 2: pentose-amino-oxazolines



M. W. Powner, B. Gerland, J. D. Sutherland, Nature 2009, 459, 239–242

# Cyanosulfidic chemistry

Interconversion of pentose aminooxazoline stereoisomers.



The "side product" ribo-21 can be converted to the expected arabino-12 by general acid-base catalysis in phosphate buffer, although with some decomposition to 26

## Cytosine-2',3'cP – step 2: pentose-amino-oxazolines



## Cytosine-2',3'cP – step 3: arabinose-anhydronucleosides



## Cytosine-2',3'cP – step 3: arabinose-anhydronucleosides



#### M. W. Powner, B. Gerland, J. D. Sutherland, *Nature* **2009**, *459*, 239–242

## **Cytosine-2',3'cP – step 4**: β-ribocytidine-2',3'-cyclic phosphate



M. W. Powner, B. Gerland, J. D. Sutherland, *Nature* **2009**, *459*, 239–242

#### **Cytosine-2',3'cP – step 4:** $\beta$ -ribocytidine-2',3'-cyclic phosphate

Rearrangement of **31**, the 3'-phosphate of *arabino*-**13**, to **1** by intramolecular nucleophilic substitution.



M. W. Powner, B. Gerland, J. D. Sutherland, Nature 2009, 459, 239–242

# Activated pyrimidine nucleotides – step 5: UV-light induced rearrangement $C \rightarrow U + cleanup$ of the side products



Orgel reported photoanomerization of alpha-cytidine **19** to the beta-anomer **25**, but the reaction was very low-yielding and combined with massive decomposition

R. A. Sanchez, L. E. Orgel, J. Mol. Biol. 1970, 47, 531-543

UV light tends to destroy most of phosphorylated cytidine derivatives (by photohydration and followed decomposition). EXCEPTION: beta-cytidine-2',3'-cyclic phosphate 1 → Partial conversion to the uridine nucleotide 33, but

both are stable on further irradiation

Photochemistry of cytidine nucleosides and nucleotides.

# **step 5:** UV-light induced rearrangement $C \rightarrow U + cleanup$ of the side products



step 5: UV-light induced rearrangement  $C \rightarrow U + cleanup$  of the side products



Photochemistry of *beta*-ribocytidine-2',3'-cyclic phosphate **1**. Under conditions of irradiation that destroy most other pyrimidine nucleosides and nucleotides, **1** undergoes partial hydrolysis and slight nucleobase loss. Ura, N1-linked uracil; Cyt–H, cytosine; Ura–H, uracil.

# Prebiotic synthesis of activated pyrimidine nucleotides

Catalysis, and reaction control through pH and chemical buffering, is indicated by dashed lines.



#### **Prebiotic route to pyrimidine nucleotides**





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